SYLLABUS - CHEM 224 – ACCELERATED

Organic Chemistry B – 2nd semester

Summer 2015 - LOYOLA UNIVERSITY CHICAGO

Lecture/Discussion: #1082 CHEM 224-001 M+W+F: 1:00 p.m. – 3:40 p.m. *** Dumbach 231 / Flanner 007 #1081 CHEM 224-002 M+W+F: 8:30 a.m. – 11:10 a.m. *** Dumbach 231 / Flanner 007 N.B. Room change on July 13th

Sr. Lecturer: Dr. C. Szpunar Office: Flanner Hall **200B** Contact: in person (preferred), 773-508-3128, cszpuna@luc.edu Emergency Message via Chemistry Dept. Office: 773-508-3100, fax: 773-508-3086 Student Office Hours: 4-5 pm Flanner 200B and directly after the morning class N.B.: Answer keys will be posted in the glass case outside Flanner 200B. No photographs!

Required:

- 1. <u>Organic Chemistry</u>, Wade 8th ed. (**red**), Prentice Hall, 2013 (ISBN 978-0-321-76841-4) *or* Wade, 7th ed. (**navy blue**), Prentice Hall, 2010 (ISBN 978-0-321-59231-6) *or* 6th ed. (**black**), Prentice Hall, 2003 (ISBN 0-13-147871-0) *or* 5th ed. (white)
- Study Guide and Solutions Manual, Wade & Simek, 8th ed. (ISBN 978-0-321-77389-0) *or* 7th ed. (ISBN 978-0-321-59871-4) *or* 6th ed. (ISBN 0-13-147882-6) *or* 5th ed., appropriate to accompany text

Suggested / Recommended Materials:

- 1. Molecular modeling kit, Darling, Prentice-Hall, Freeman (Maruzen), Proteus, or equivalent
- 2. Spiral or bound notebook for homework problems

***NO ACCESS CODE NEEDED, NOR DESIRED, NOR SUGGESTED

Optional Materials (found helpful by some students):

- 1. Organic Chemistry as a Second Language, II, Klein (2006), Wiley (ISBN 978-0-471-73808-4)
- 2. <u>Barron's Orgo Cards</u>: Organic Chemistry Review, Wang, Razani, Lee, Wu, and Berkowitz (ISBN 0-7641-7503-3)

Grading (weighting below) with approximate curved grade guidelines: > 90% A; 75-90% B; 55-75% C

MID-TERM EXAM – date scheduled and announced (subject to change, although unlikely)

30%

III NO MAKE UPS III

- UNEXCUSED ABSENCES merit a zero score.
- EXCUSED ABSENCES are handled on a case-by-case basis; grade weighting may be adjusted, depending on the circumstance(s); however, an excused absence **MUST BE CORROBORATED and DOCUMENTED**, e.g., accompanied by a note from the doctor, dentist, hospital rep, or funeral director; by a court summons, plane ticket stub, hospital release form, obituary, or other. With proper documentation, religious observance, official representation of the university, or personal emergency may constitute an Excused Absence.

J OUIZZES – 4 – dates announced (subject to change, although unlikely), NO MAKE UPS !!! 30%

J J FINAL – date announced (scheduled by CAS), no alternative date/time, NO MAKE UPS !!! 40%

II II II Homework - assigned per chapter; feel free to work any/all problems to apply and master concepts.

*** Please note that because this course, *Organic Chemistry, is cumulative, comprehensive, and improvement-based,* and because the final exam is deemed a culminating measure of a student's progress, any student meriting an F on the final exam may achieve a recorded course grade no higher than D, despite total points; a final-exam score of D may merit a course grade no higher than C, despite total points; and a final-exam score of C may merit a course grade no higher than B, despite student's standing otherwise (i.e., despite total points.) **Course Objective**: To guide, encourage, and foster the learning and understanding of Organic Chemistry – nomenclature, structures, properties, mechanisms, syntheses, and spectroscopy – by the individual student, helping him/her to connect, extrapolate, integrate, and apply the many different aspects learned.

Student Outcomes: If successful, the student will learn how to ...

- 1. identify the various classes of organic compounds, their methods of preparation, and typical reactions.
- 2. name and draw specific organic compounds.
- 3. postulate a logical reaction mechanism for simple organic reactions.
- 4. discriminate amongst relative stabilities of reaction intermediates.
- 5. plan and write out multi-step syntheses using known reagents / conditions to transform functional groups.
- prepare for basic purification/separation techniques of organic compounds required in the laboratory.
 analyze and interpret data from various instruments used in separating and identifying organic compounds: IR, NMR, and UV-vis spectrophotometers and mass spectrograph.

Lecture and Discussion – Attendance and Attention: Important and required. Feel free to bring your books and modeling kit to class. Better yet, use them. Prepare for lecture by prior scanning of new material. Come prepared for discussion; be ready to ask questions on assigned homework or yet-unassimilated lecture material.

Cell Phones: NONE. Please be courteous and respectful of others. Silent mode during lecture and discussion. *Not allowed in sight or within hearing during exams, subject to confiscation.* NO phone conversations in lecture hall or in discussion class – before class, during class, after class – AT ANY TIME! NO texting – before class, during class, after class – AT ANY TIME! If you must talk or text, take it outside!!!

Photography: NONE. No photography of posted quiz/exam keys. No photography of discussion/lecture blackboard / whiteboard.

Recording: NONE. No recording of lectures.

Academic Honesty: Essential, expected, and enforced. Dishonesty dictates consequences which may include: (1) notification of Chemistry Department Chair, student's Department Chair, and CAS Dean, (2) documentation in the student's official university record, and (3) dismissal from the university. <u>Immediate consequences</u> will include a **ZERO** on any item in question (quiz or exam). Please refer to the LUC Undergraduate Handbook on policies or the CAS website: <u>http://www.luc.edu/academics/catalog/undergrad/reg_academicintegrity.shtml</u>.

Study Strategies and Suggestions: One may approach the study of Organic Chemistry in a manner similar to tackling a new foreign language. Its study will provide a basis to understanding future material – *building constantly, incessantly, and relentlessly* on the structural and mechanistic information presented previously and, hopefully, acquired by the student. Over two semesters, the course will cover: bonding, functional groups, families of aliphatic and aromatic compounds, nomenclature, structures, stereochemistry, reaction mechanisms, multi-step syntheses, and spectroscopic techniques. Because the course is cumulative and builds heavily on prior material, the best plan is to study Organic Chemistry regularly, every day, similar to practicing the piano. Collaboration on homework problems is encouraged, especially in a timely fashion. Experience dictates that positive outcomes (for exam and course grades) are directly proportional to working and understanding the assigned problems on a regular basis, i.e., applying the concepts learned to non-generic situations.

Typically, Organic Chemistry is not efficiently self-taught. <u>Overnight cramming will probably not produce success</u>! The student should quickly read the chapter/segment to be covered BEFORE lecture to improve lecture comprehension. After lecture, careful detailed re-reading of the chapter/segment and focused <u>working of the assigned problems are appropriate</u>, <u>necessary</u>, <u>essential</u>, <u>and expected</u>. In addition to student's participation in lecture, discussion, reading, and homework, joining and contributing to a study group is strongly encouraged.

If anticipating a passing grade of **C**, the minimal time per week <u>in the summer</u> devoted to Organic Chemistry is estimated at 9 hr for lecture/discussion, 6-12 hr for reading, and 6-12 hr for homework.

Suggested Homework Assignment (for Wade's [red] 8th edition):

Chap 15: 1, 4-18, 22-27, 30, 33 Chap 16: 3-4, 7-8, 9 (a,b), 12-29, 31-32, 38-39, 45 Chap 17: 1-29, 33-35, 37-40, 42-44, 46-52, 54 Chap 18: 1-4, 6-11, 15-30, 33-39, 43-44, 46-51 Chap 19: 1-21, 25-33, 34-37, 39, 42, 44, 47, 51, 53 Chap 20: 1-21, 23-33, 35-40, 45 Chap 21: 1-39, 42-54, 62, 64-65 Chap 22: 1-47 Chap 23: 1-14, 16-17, 21-22, 24-26, 28-31, 32 (a), 33-36, 40, 41 (a,b), 52-53, 55, 58 Chap 24: 1-3, 20, 33 Chap 25: 1-5, 9-13, 15, 30 Chap 26: 17 and as warranted

Suggested Homework Assignment (Wade 7th edition, [navy blue]):

Chap 15: 1, 4-18, 22-27, 30 Chap 16: 3-4, 7-8, 9 (a,b), 12-29, 32, 38-39, 45 Chap 17: 1-2, 4-22, 24-27, 30-32, 38, 40-52 Chap 18: 1-4, 6-12, 16-31, 34-40, 43-44, 47, 49, 51 Chap 19: 1-21, 25-32, 34-42, 44, 47, 56, 58 Chap 20: 1-33, 35-40, 45, 47, 50 Chap 21: 1-39, 43-48, 50-54, 66 Chap 22: 1-47, 59-65, 67-69 Chap 23: 1-14, 16-17, 21-22, 24-26, 28-31, 32 (a), 33-36, 40, 41 (a,b), 52-55, 63 Chap 24: 3-6, 20, 32, 33 Chap 25: 1, 4, 8-9, 11-15, 32 Chap 26: FYI, 21-29

Suggested Homework Assignment (Wade 6th edition, [black])

Chap 15: 1, 4-18, 22-27, 30 Chap 16: 3-4, 7-8, 9 (a,b), 12-29, 32, 38-39, 45 Chap 17: 1-2, 4-22, 24-27, 30-32, 38, 40-52 Chap 18: 1-4, 6-12, 16-31, 34-40, 43-44, 47, 49, 51 Chap 19: 1-21, 25-32, 34-42, 44, 47, 56, 58 Chap 20: 1-33, 35-40, 45, 47, 50 Chap 21: 1-39, 43-48, 50-54, 66 Chap 22: 1-47, 59-65, 67-69 Chap 23: 1-14, 16-17, 21-22, 24-26, 28-31, 32 (a), 33-36, 40, 41 (a,b), 52-55, 63 Chap 25: 1, 4, 8-9, 11-15, 32 Chap 26: 21-29

12/13. Spectroscopy: Methods of Structure Determination (Review)

Electromagnetic spectrum, molecular vibrations, infrared spectroscopy, characteristic IR absorptions, IR spectra interpretation, mass spectrometry (MS), terminology associated with MS, nuclear spin, magnetic shielding, NMR spectrometer, chemical shift, chemical nonequivalence, peak integration, spin-spin splitting, time dependence, interpreting NMR spectra, combined spectroscopy problems.

15. Dienes, Conjugated Systems, UV Spectroscopy

Molecular orbital theory, 1,3-butadienes, electrophilic addition (1,2 and 1,4), NBS, Diels Alder rxn.

16/17. Aromatic Compounds

Benzene, Kekule structure, resonance, annulenes, MO theory, aromaticity, Huckel's Rule, heterocyclic aromatics, electrophilic aromatic substitution, directing effects on electrophilic aromatic substitution, Friedel-Crafts alkylation and acylation, nucleophilic aromatic substitution, side-chain reactions of benzene derivatives, reactions of phenols.

18. Aldehydes and Ketones

Structure of the carbonyl group; nomenclature, physical properties, and spectroscopic properties of aldehydes and ketones; syntheses of aldehydes and ketones; addition reactions; Wittig reaction; condensation reactions; acetals; oxidation and reduction.

19. Amines

Nomenclature, structure, physical properties, basicity, salts of amines, phase transfer catalysis, spectroscopic properties, reactions of amines, sulfonamides, Hofmann elimination, Cope Elimination, arenediazonium salts, synthesis of amines.

20. Carboxylic Acids

Structure and nomenclature, physical properties, acidity, salts of carboxylic acids, spectroscopy, synthesis of carboxylic acids, nucleophilic acyl substitution, Fischer esterification, synthesis and use of acid chlorides, diazomethane, condensations of carboxylic acids, reduction, alkylation of carboxylic acids.

21. Carboxylic Acid Derivatives

Structure and nomenclature, physical properties, and spectroscopic properties of carboxylic acid derivatives, nucleophilic acyl substitution, hydrolysis of acid derivatives, reduction of acid derivatives, organometallic reactions, thioesters, carbonic acid derivatives.

22. Enols and Enolates

Enols, enolates, alpha halogenation, alkylation of enolates, formation and alkylation enamines, aldol condensation, dehydration of aldol products, crossed aldol condensation, aldol cyclizations, Claisen condensation, Dieckmann condensation, crossed Claisen condensation, β -dicarbonyl compounds, Malonic ester synthesis, acetoacetic ester synthesis, conjugate addition, Michael reaction, Robinson annulation.

23. Carbohydrates

Classification of carbohydrates, monosaccharides, erthyro and threo diastereromers, epimers, nomenclature, cyclic structures of monosaccharides, anomers, mutarotation, reactions of monosaccharides, oxidation and reduction of monosaccharides, non-reducing sugars, formation of glycosides, ether and ester formation, osazone formation, Ruff degradation, Kiliani-Fischer synthesis, Fischer's proof of configuration, determination of ring size, periodic acid cleavage, disaccharides, polysaccharides.

24. Amino Acids and Peptides

Structure and stereochemistry of the α -amino acids, synthesis of amino acids, resolution, structure and nomenclature of peptides and proteins, peptide structure determination, peptide synthesis.

25. Lipids

Miscellaneous category, classification by solubility: waxes, triglycerides, soaps and detergents, phospholipids, steroids, prostaglandins, terpenes.

Lecture Outline (tentative, subject to change, but unlikely due to time constraints)

Lecture	<u>Date</u>	<u>Chapters</u>	Topic	*** EVENT ***
1 2 3	M-June 29 W-July 1 F-July 3	12/13 12/13/15 ***	Review – IR, MS, NMR / Conjugated Systems, UV Spectro ************************************	
4 5 6	M-July 6 W-July 8 F-July 10	16 16 17	Aromatic Compounds - Reactions	z 1 DUE (Chapters 12/13) Quiz 2 (Chapters 15-17)
7 8 9	M-July 13 W-July 15 F-July 17	17 17/18 18	 / Aldehydes and Ketones ********** MID-TERM EXAM (Cha	apters 12-13, 15-18 partial)
10 11 12	M-July 20 W-July 22 F-July 24	18 19 19	 Amines *********************************	Quiz 3 (Chapters 18-19)
13 14 15	M-July 27 W-July 29 F-July 31	19/20 20 21	/ Carboxylic Acids Carboxylic Acid Derivatives ************************************	Quiz 4 (Chapters 20-21) W, not WF ****************
16 17 18	M-Aug 3 W-Aug 5 F-Aug 7	22 23/24 25	Condensations, Alpha Substitutions, Carbohydrates / Amino Acids and Pe Lipids / Q/A ********* Cumulative FINAL EXAM	eptides

Daily Schedule - Mornings (tentative, approximate, flexible, may adjust order):

<u>Regular Day</u>	<u>Quiz Day</u>	<u>Exam Day</u>
08:30 – 09:00 am Q/A, admin 09:00 – 09:50 lecture – 1	08:30 – 09:00 am Q/A 09:00 – 09:50 lecture - 1	08:30 – 09:00 Q/A 09:00 – 10:10 lecture
09:50 - 10:00 ***break***	09:50 - 10:00 ***break***	10:10 – 10:20 ***break***
10:00 – 10:20 discussion as time/topic permit	10:00 – 10:50 lecture – 2	10:20 – 11:10 <i>EXAM</i>
10:20 – 11:10 lecture - 2	10:50 – 11:10 <i>quiz</i>	******
		08:30 – 09:00 Q/A 09:00 – 09:10 ***break*** 09:10 – 11:10 <i>FINAL</i>

Daily Schedule – <u>Afternoons</u> (tentative, approximate, flexible, subject to adjustment):

<u>Regular Day</u>	<u>Quiz Day</u>	Exam Day
01:00 – 01:30 pm Q/A, admin 01:30 – 02:20 lecture – 1	01:00 – 01:30 pm Q/A 01:30 – 02:20 lecture - 1	01:00 – 01:30 pm Q/A 01:30 – 02:40 lecture
02:20 - 02:30 ***break***	02:20 - 02:30 ***break***	02:40 - 02:50 ***break***
02:30 – 02:50 discussion as time/topic permit	02:30 – 03:20 lecture – 2	02:50 – 03:40 <i>EXAM</i>
02:50 - 03:40 lecture - 2	03:20 – 03:40 <i>quiz</i>	*****
		01:00 – 01:30 Q/A

01:00 – 01:30 Q/A 01:30 – 01:40 ***break*** 01:40 – 03:40 *FINAL*